

Perfluoro/polyfluoroalkoxysulfonyl fluorides – precursors to new ion conductors, fuel cell electrolytes and polymeric coatings

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1. Introduction

Perfluoro/polyfluoroalkoxysulfonyl fluorides continue to be subjects of considerable interest because of their utility in preparing advanced materials containing the sulfonyl fluoride grouping ($=\text{SO}_2\text{F}$) or its derivatives.

2. Results and discussion

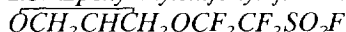
2.1. Perfluoroalkanesulfonyl fluorides, $\text{FSO}_2(\text{CF}_2)_n\text{SO}_2\text{F}$

These compounds were converted to their corresponding sulfonic acids, $\text{HSO}_3(\text{CF}_2)_n\text{SO}_3\text{H}$ with $n=2, 4, 6$, and lithium salts, $\text{LiSO}_3(\text{CF}_2)_n\text{SO}_3\text{Li}$ with $n=1-4$. Evaluation of the acids as fuel cell electrolytes [1] and of the lithium salts as ion conductors have been carried out [2].

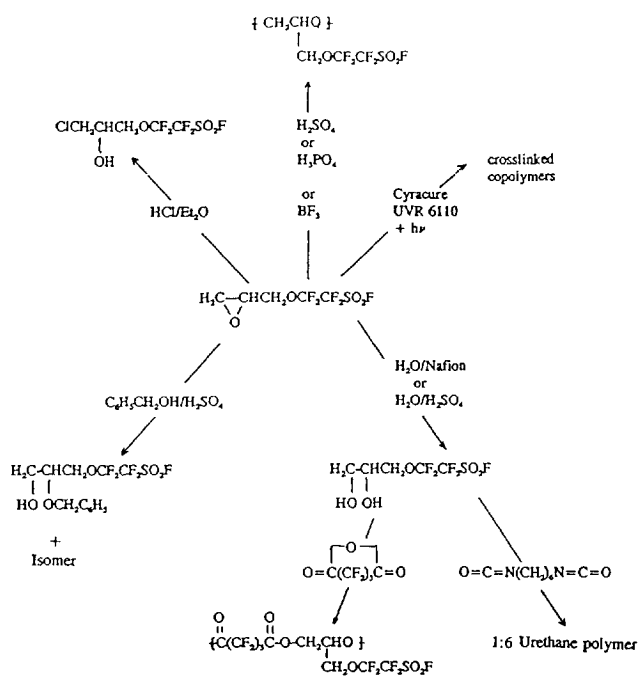
2.2. SF_5 sultones, $\text{SF}_5\text{C}(\text{XCF}_2)\text{OSO}_2$ ($X=H, F$)

SF_5 -containing sultones were used to prepare the following SF_5 -containing acids: $\text{SF}_5\text{CH}_2\text{SO}_3\text{H}$, $\text{SF}_5\text{CHF}_2\text{SO}_3\text{H}$ and $\text{SF}_5\text{CF}_2\text{SO}_3\text{H}$. The possible use of these acids as fuel cell electrolytes looks promising. Lithium salt complexes over the stoichiometric range $(\text{CH}_2\text{CH}_2\text{O})_x\text{LiA}$ ($A=\text{SF}_5\text{CFH}_2\text{SO}_3, \text{SF}_5\text{CF}_2\text{SO}_3$) with $x=4-16$ have been prepared and evaluated [2]. Surprisingly, these complexes showed conductivities in the range reported for complexes of polyethylene oxide with $\text{LiN}(\text{SO}_3\text{CF}_3)_2$ and $\text{LiC}(\text{SO}_3\text{CF}_3)_3$.

2.3. Epoxyalkylsulfonyl fluoride,



This epoxide was used to prepare a number of interesting derivatives as shown in Scheme 1 [3,4]. Photocured copolymer films containing various amounts of FSO_2 epoxide monomer and a commercial cycloaliphatic diepoxide were analyzed by X-ray photoelectron spectroscopy and secondary ion mass spectrometry. Preliminary results indicated significant fluorine concentrations in the outer surface of the film [5].



Scheme 1. Compounds derived from epoxyalkylsulfonyl fluoride, $\text{OCH}_2\text{CHCH}_2\text{OCF}_2\text{CF}_2\text{SO}_2\text{F}$.

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References

- [1] H. Saffarian, P. Ross, F. Behr and G.L. Gard, *J. Electrochem. Soc.*, 139 (1992) 2391.
- [2] In-house work at Portland State University and Oregon State University.
- [3] N.N. Hamel, G.A. Russell and G.L. Gard, *J. Fluorine Chem.*, 66 (1994) 105.
- [4] N.N. Hamel and G.L. Gard, *J. Fluorine Chem.*, 68 (1994) 253.
- [5] D.D. Castner, D.N. Grainger and G.L. Gard, personal communication, 1994.